

## Ozone Production and it's Sensitivity to NO<sub>x</sub> and VOCs

Only an accurate chemical transport model working off of an accurate emission inventory can provide a full answer to the "what-if" question of what happens when emissions are changed. A complimentary approach which is independent of an emission inventory is to use observations to constrain a box model and calculate the rate of O<sub>3</sub> production, P(O<sub>3</sub>), and it's sensitivity to NO<sub>x</sub> and VOCs. Through the model derived variable Ln/Q (fraction of radicals removed by reactions with NO<sub>x</sub>/ total primary radical production), one can determine where the atmosphere is in the continuum between NO<sub>x</sub> limited and VOC limited conditions. We present calculated P(O<sub>3</sub>) and sensitivities based on G-1 observations taken over the Mexico City plateau. The highest P(O<sub>3</sub>)'s occur under VOC limited conditions suggesting that peak O<sub>3</sub> itself is VOC limited. Even though VOC/NO<sub>x</sub> ratios are higher than observed in U.S. cities (which by itself tilts the atmosphere towards being NO<sub>x</sub> limited), VOC limited conditions occur because of extremely high NO<sub>x</sub> concentrations. Most of the time to time and point to point variability in P(O<sub>3</sub>) under these VOC limited conditions is due to variations in radical production rate. Under VOC limited conditions NO<sub>z</sub> rather than peroxides is an end product of photochemistry. Out of 50 O<sub>3</sub> plumes observed from the G-1, only 2 show a correlated H<sub>2</sub>O<sub>2</sub> peak, which is consistent with VOC limited conditions.

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